MATLAB for Engineers, 6th Edition

Chapter 2 Homework Solutions

```
clear,clc, format shortg
```

You can either solve these problems in the command window, using MATLAB® as an electronic calculator, or you can create a program using a script (M-file) or a live script (MLX-file). If you are solving these problems as a homework assignment or if you want to keep a record of your work, the best strategy is to use a program file, divided into sections with section dividers.

```
clear, clc
```

Getting Started

Problem 2.1

Predict the outcome of the following MATLAB® calculations. Check your results by entering the calculations into the command window.

```
1 + 3/4
               1.75
ans =
5*6*4/2
ans =
          60
5/2*6*4
          60
ans =
5^2*3
ans =
          75
5^(2*3)
             15625
ans =
1 + 3 + 5/5 + 3 + 1
           9
ans =
(1 + 3 + 5)/(5 + 3 + 1)
ans =
```

Using Variables

Identify which name in each of the following pairs is a legitimate MATLAB® variable name. Test your answers by using isvarname—for example,

isvarname fred

The function isvarname returns a 1 if the name is valid and a 0 if it is not. Although it is possible to reassign a function name as a variable name, doing so is not a good idea. Use which to check whether the preceding names are function names—for example,

which sin

In what case would MATLAB® tell you that sin is a variable name, not a function name?

The legitimate Matlab names are: fred book 1 Second Place No 1 vel 5 tan

```
isvarname fred
 ans = logical
    1
isvarname book_1
 ans = logical
    1
isvarname Second_Place
 ans = logical
    1
isvarname No_1
 ans = logical
    1
isvarname vel_5
 ans = logical
    1
                  %although tan is a function name it can be used as a variable
isvarname tan
name
 ans = logical
    1
isvarname fred! %! is not an allowed character
 ans = logical
    0
```

```
isvarname book-1 % - is not an allowed character
ans = logical
   0
isvarname 2ndplace % variable names must start with a letter
ans = logical
isvarname #1 % # is not an allowed character
ans = logical
   0
isvarname vel.5 % . is not an allowed character
ans = logical
   0
isvarname while % while is a reserved name
ans = logical
   0
which tan % tan is a function name
built-in (C:\Program Files\MATLAB\R2022a\toolbox\matlab\elfun\@double\tan) % double method
which while % while is also a function name, but is reserved
built-in (C:\Program Files\MATLAB\R2022a\toolbox\matlab\lang\while)
```

You can reassign a function name as a variable name

For example:

```
sin = 3
sin = 3
```

The which function now tells us sin is a variable

```
which sin
sin is a variable.
```

Use the clear function to return sin to its function definition

```
clear sin
which sin
```

built-in (C:\Program Files\MATLAB\R2022a\toolbox\matlab\elfun\@double\sin) % double method

Scalar Operations and Order of Operations

Problem 2.3

Create MATLAB® code to perform the following calculations. Check your code by entering it into MATLAB® and performing the calculations on your scientific calculator.

5^2 ans = 25 (5 + 3)/(5*6)0.26667 ans = sqrt(4 + 6^3) % or... ans = 14.832 $(4 + 6^3)^(1/2)$ ans = 14.832 $9+6/12 + 7*5^{(3+2)}$ ans = 21884 $1 + 5*3/6^2 + 2^2(2-4) *1/5.5$ 1.4621

Problem 2.4

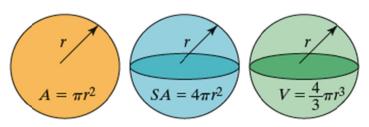


Figure 2.4

- (a) Area of a circle, (b) Surface area of a sphere,
- (c) Volume of a sphere with radius r.

(a) The area of a circle is πr^2 . Define r as 5, then find the area of a circle, using MATLAB®.

$$r = 5$$

$$r = 5$$

$$area = pi*r^2$$

(b) The surface area of a sphere is $4\pi r^2$. Find the surface area of a sphere with a radius of 10 ft.

```
r = 10

r = 10

surface_area = 4*pi*r^2

surface area = 1256.6
```

(c) The volume of a sphere is $\frac{4}{3}\pi r^3$. Find the volume of a sphere with a radius of 2 ft.

```
r = 2

r = 2

volume = 4/3*pi*r^3

volume = 33.51
```

Problem 2.5

area =

78.54

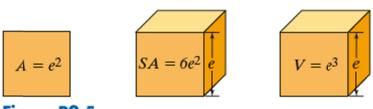


Figure P2.5

(a) Area of a square, (b) Surface area of a cube, (c) Volume of a cube with edge length e.

(a) The area of a square is the edge length squared ($A = edge^2$). Define the edge length as 5, then find the area of a square, using MATLAB®.

```
edge = 5
edge = 5

area = edge^2
area = 25
```

(b) The surface area of a cube is 6 times the edge length squared

(SA = $6 \times edge^2$). Find the surface area of a cube with edge length 10.

```
edge = 10

edge = 10

surface_area = 6*edge^2

surface_area = 600
```

(c) The volume of a cube is the edge length cubed ($V = edge^3$). Find the volume of a cube with edge length 12.

```
edge = 12

edge = 12

volume = edge^3

volume = 1728
```

Problem 2.6



Figure P2.6

The geometry of a barbell can be modeled as two spheres and a cylindrical rod.

Consider the barbell shown in Figure P2.6.

(a) Find the volume of the figure, if the radius of each sphere is 10 cm, the length of the bar connecting them is 15 cm, and the diameter of the bar is 1 cm. Assume that the bar is a simple cylinder.

```
r=10; %cm
length=15; %cm
d=1; % cm
% Find the volume of each sphere
volume_sphere=4/3*pi*r^3;
% Find the volume of the bar
volume_bar=pi*(d/2)^2*length;
% Combine the components to get the total volume
total_volume=2*volume_sphere +volume_bar
```

b)Surface Area

total volume =

(b) Find the surface area of the figure.

8389.4

Find the surface area of each sphere

```
sa_sphere=4*pi*r^2;
% Find the surface area of the bar
sa_bar=pi*d*length;
% Combine the components to get the total surface area
total_sa=2*sa_sphere + sa_bar

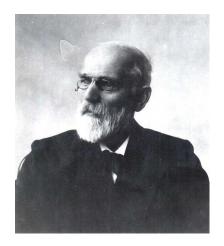
total_sa = 2560.4
```

Problem 2.7

The ideal gas law was introduced in Example 2.1. It describes the relationship between pressure (P), temperature (T), volume (V), and the number of moles of gas (n).

PV = nRT

The additional symbol, *R*, represents the ideal gas constant. The ideal gas law is a good approximation of the behavior of gases when the pressure is low and the temperature is high. (What constitutes low pressure and high temperature varies with different gases.) In 1873, Johannes Diderik van der Waals, (see Figure P2.7) proposed a modified version of the ideal gas law that better models the behavior of real gases over a wider range of temperature and pressure.



$$\left(\left(P + \frac{n^2 a}{V^2}\right)(V - nb)\right) = nRT$$

In this equation the additional variables *a* and *b* represent values characteristic of individual gases.

Use both the ideal gas law and van der Waals' equation to calculate the temperature of water vapor (steam), given the following data.

```
P=220; % Pressure in bars
n=2; % Moles
V=1; % Volume in liters
a=5.536; % constant in L^2 bar/mol^2
b=0.03049; % constant in L/mol
R=0.08314472; % ideal gas constant L bar/(K mole)
```

Find the temperature using the ideal gas law

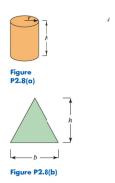
```
T_ideal=P*V/(n*R) % kelvins

T_ideal = 1323
```

Find the temperature using Van der Waal's equation

$$T_VW=(P + n^2*a/V^2)*(V - n*b)/(n*R)$$

 $T_VW= 1367.4$



(a) The volume of a cylinder is $V = \pi r^2 h$. Define r as 3 and h as the array

$$h = [1, 5, 12]$$

Find the volume of the cylinders (see Figure P2.8a).

```
r=3;
h=[1,5,12];
volume = pi*r^2.*h
volume = 1×3
28.274 141.37 339.29
```

Note that we need to use the .* operator because h is an array

(b) The area of a triangle is 1/2 the length of the base of the triangle, times the height of the triangle. Define the base as the array

$$b = [2, 4, 6]$$

and the height h as 12, and find the area of the triangles

```
b=[ 2, 4, 6];
h=12;
area=1/2*b.*h
area = 1×3
12 24 36
```

Although you don't have to use the .* operator for both multiplications and the ./ for the division it won't hurt if you do and is good practice.

```
area=1./2.*b.*h

area = 1×3

12 24 36
```

(c) The volume of any right prism is the area of the base of the prism, times the vertical dimension of the prism. The base of the prism can be any shape—for example, a circle, a rectangle, or a triangle. Find the volume of the prisms created from the triangles of part (b). Assume that the vertical dimension of these prisms is 6

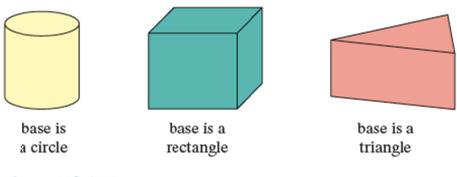


Figure P2.8(c)

```
h=6;
volume=h.*area
```

volume = 1×3 72 144 216

Problem 2.9

The response of circuits containing resistors, inductors and capacitors depends upon the relative values of the resistors and the way they are connected. An important intermediate quantity used in describing the response of such circuits is *s*. Depending on the values of *R*, *L*, and *C*, the values of *s* will be either both real values, a pair of complex values, or a duplicated value.

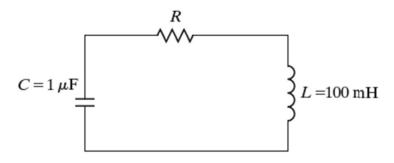


Figure P2.9

Series circuit.

The equation that identifies the response of a particular series circuit (Figure P2.9) is

$$S = \frac{-R}{2L} \pm \sqrt{\left(\frac{R}{2L}\right)^2 - \frac{1}{LC}}$$

(a) Determine the values of s for a resistance of 800 Ω .

```
R=800 %ohms
       800
R =
L=100e-3 %H
L =
              0.1
C=1e-6 %F
            1e-06
C =
s(1) = -R/(2*L) + sqrt((R/(2*L))^2 - 1/(L*C))
          -1550.5
s =
s(2) = -R/(2*L) - sqrt((R/(2*L))^2 - 1/(L*C))
s = 1 \times 2
        -1550.5
                       -6449.5
```

(b) Create a vector of values for R ranging from 100 to 1000 Ω and evaluate s. Refine your values of R until you find the approximate size of resistor that yields a pure real value of s. Describe the effect on s as R increases in value.

```
Hint:
1 μF = 1e-6 F
1 mH = 1e-3 H
```

500 +

```
1 \text{ mH} = 1e-3 \text{ H}
 R=100:100:1000; % array of R values
 s_{plus} = -R./(2*L) + sqrt((R./(2*L)).^2 - 1/(L*C))
  s_plus = 1 \times 10 \text{ complex}
             -500 + 3122.5i
                                          -1000 +
                                                           3000i ···
 s minus = -R./(2*L)-sqrt((R./(2*L)).^2 - 1/(L*C))
  s minus = 1 \times 10 complex
             -500 -
                      3122.5i
                                          -1000 -
                                                           3000i ···
 results = [R',s_plus',s_minus']
  table values = 10 \times 3 complex
                                                      3122.5i ...
              100 +
                              Οi
                                           -500 -
              200 +
                             Οi
                                           -1000 -
                                           -1500 - 2783.9i
-2000 - 2449.5i
              300 +
                             Oi
              400 +
                             Οi
```

-2500 -

1936.5i

Οi

```
600 +
                                                     1000i
                           Οi
                                      -3000 -
                       0i
0i
           700 +
                                      -2000 +
                                                        Οi
           + 008
                                    -1550.5 +
                                                        Οi
                         Oi
                                                        Οi
           900 +
                                    -1298.4 +
          1000 +
                         Οi
                                      -1127 +
                                                        Οi
% Repeat for R between 600 and 700
R=600:10:700;
s_{plus} = -R./(2*L) + sqrt((R./(2*L)).^2 - 1/(L*C))
s_plus = 1 \times 11 complex
                        1000i
                                      -3050 +
         -3000 +
                                                   835.16i ···
s minus = -R./(2*L)-sqrt((R./(2*L)).^2 - 1/(L*C))
s_minus = 1 \times 11 complex
         -3000 -
                        1000i
                                      -3050 -
                                                   835.16i ···
results = [R',s plus',s minus']
table values = 11×3 complex
           600 +
                           Οi
                                      -3000 -
                                                     1000i ···
           610 +
                         Οi
                                      -3050 -
                                                  835.16i
           620 +
                         Οi
                                     -3100 -
                                                  624.5i
                         Οi
                                      -3150 -
                                                   278.39i
           630 +
                       0i
0i
0i
0i
0i
           640 +
                                    -2710.1 +
                                                        Οi
           650 +
                                      -2500 +
                                                        Οi
           660 +
                                   -2356.6 +
                                                        Οi
           670 +
                                    -2244.3 +
                                                        Οi
           680 +
                                     -2151 +
                                                        Οi
                     Oi
           690 +
                                    -2070.7 +
                                                        Οi
```

The equation that identifies the response parameter, s, of the parallel circuit shown in Figure P2.10

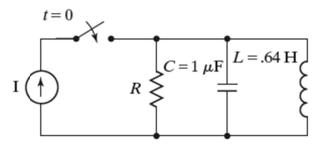


Figure P2.10

Parallel circuit.

$$S = -\frac{1}{2RC} \pm \sqrt{\left(\frac{1}{2RC}\right)^2 - \frac{1}{LC}}$$

(a) Determine the values of s for a resistance of 200 Ω .

```
R=200;

C=1e-6;

L=0.64

s_plus=-1/(2*R*C) + sqrt((1/(2*R*C))^2 - 1/(L*C))

s_plus = -334.94

s_minus=-1/(2*R*C) - sqrt((1/(2*R*C))^2 - 1/(L*C))

s_minus = -4665.1
```

(b) Create a vector of values for R ranging from 100 to 1000 Ω and evaluate s. Refine your values of R until you find the size of resistor that yields a pure real value of s. Describe the effect on s as R decreases.

```
R=100:100:1000;
s_{plus}=-1./(2*R*C) + sqrt((1./(2*R*C)).^2 - 1/(L*C));
s_{minus}=-1./(2*R*C) - sqrt((1./(2*R*C)).^2 - 1/(L*C));
results = [R',s plus',s minus']
results = 10×3 complex
                         0i -158.77 + 0i

0i -334.94 + 0i

0i -564.27 + 0i

0i -1250 + 0i

0i -1000 - 750i

0i -833.33 - 931.69i

0i -714.29 - 1025.8i

0i -625 - 1082.5i

0i -555.56 - 1119.8i

0i -500 - 1145.6i
                100 +
                                                                              0i …
                200 +
                300 +
                500 +
600
                400 +
                600 +
                700 +
                800 +
                900 +
               1000 +
```

Refine the R values

```
-1190.5 - 381.14i

-1162.8 - 458.71i

-1136.4 - 520.75i

-1111.1 - 572.65i

-1087 - 617.27i
420 +
                     Οi
               01
0i
0i
0i
0i
0i
0i
430 +
440 +
                              -1136.4 -
-1111.1 -
450 +
460 +
                              -1063.8 -
470 +
                                                   656.33i
480 +
                              -1041.7 -
                                                   690.96i
490 +
                               -1020.4 -
                                                   721.99i
```

Burning one gallon of gasoline in your car produces 19.4 pounds of CO2. Calculate the amount of CO2 emitted during a year for the following vehicles, assuming they all travel 12,000 miles per year. The reported fuel-efficiency numbers were extracted from the U.S. Department of Energy website, www.fueleconomy.gov, and reflect the combined city and highway estimates.

Create an array of mpg values

```
mpg=[142, 36, 52, 35, 41,113];
```

Calculate the emissions

```
Mass_CO2=(12000./mpg*19.4)'

Mass_CO2 = 6×1

1639.4
6466.7
4476.9
6651.4
5678
2060.2
```

Notice that the result was transposed so that it is easier to read.

Problem 2.12

(a) Create an evenly spaced vector of values from 1 to 20 in increments of 1.

```
a=1:20

a = 1×20

1 2 3 4 5 6 7 8 9 10 11 12

13 ···
```

(b) Create a vector of values from zero to 2π in increments of $\frac{\pi}{10}$.

```
b=0:pi/10:2*pi
b = 1×21
```

```
0 0.31416 0.62832 0.94248 1.2566
1.5708 ···
```

(c) Create a vector containing 15 values, evenly spaced between 4 and 20. (*Hint:* Use the linspace command. If you can't remember the syntax, type help linspace.)

(d) Create a vector containing 10 values, spaced logarithmically between 10 and 1000. (*Hint:* Use the logspace command.)

```
d=logspace(1,3,4)

d = 1×4

10     46.416     215.44     1000
```

Problem 2.13

(a) Create a table of conversions from feet to meters. Start the feet column at 0, increment it by 1, and end it at 10 feet. (Look up the conversion factor in a textbook or online.)

```
feet=0:1:10;
meters=feet./3.28;
[feet', meters']
ans = 11 \times 2
               \cap
                              0
               1
                      0.30488
               2
                       0.60976
               3
                       0.91463
                        1.2195
               4
               5
                        1.5244
               6
                        1.8293
              7
                        2.1341
               8
                        2.439
                        2.7439
       :
```

(b) Create a table of conversions from radians to degrees. Start the radians column at 0 and increment by 0.1π radians, up to π radians. (Look up the conversion factor in a textbook or online.)

```
radians=0:0.1*pi:pi;
degrees=radians*180/pi;
[radians',degrees'] % Or we could also give this array a name
ans = 11×2
0 0
```

```
0.31416
               18
0.62832
               36
0.94248
1.2566
               72
1.5708
              90
 1.885
              108
2.1991
              126
2.5133
              144
2.8274
              162
```

conversion_table=[radians',degrees']

```
conversion_table = 11×2
          0
                      0
     0.31416
                      18
     0.62832
                     36
     0.94248
                     54
      1.2566
                      72
      1.5708
                     90
      1.885
                    108
      2.1991
                    126
      2.5133
                    144
      2.8274
                    162
```

(c) Create a table of conversions from mi/h to ft/s. Start the mi/h column at 0 and end it at 100 mi/h. Print 15 values in your table. (Look up the conversion factor in a textbook or online.)

```
mph=linspace(0,100,15);
ft_per_sec=mph*5280/3600;
vel_conversion=[mph',ft_per_sec']
```

```
vel\_conversion = 15 \times 2
         0
                10.476
      7.1429
      14.286
                20.952
      21.429
                 31.429
      28.571
                 41.905
      35.714
                52.381
      42.857
                62.857
         50
                73.333
      57.143
                 83.81
      64.286 94.286
```

(d) The acidity of solutions is generally measured in terms of pH. The pH of a solution is defined as $-\log 10$ of the concentration of hydronium ions. Create a table of conversions from concentration of hydronium ion to pH, spaced logarithmically from .001 to .1 mol/liter with 10 values. Assuming that you have named the concentration of hydronium ions H_{conc} , the syntax for calculating the negative of the logarithm of the concentration (and thus the pH) is

```
pH = -log10(H_conc)
```

The general equation for the distance that a freely falling body has traveled (neglecting air friction) is

$$d = \frac{1}{2}gt^2$$

Assume that $g = 9.8 \frac{m}{s^2}$. Generate a table of time versus distance traveled for values of time from 0 to 100 seconds. Choose a suitable increment for your time vector. (*Hint:* Be careful to use the correct operators; t^2 is an array operation!)

```
g=9.8;
t=0:10:100;
d=1/2*g*t.^2;
results=[t',d']
table values = 11 \times 2
             0
                           0
            10
                        490
            20
                       1960
            30
                       4410
            40
                       7840
            50
                      12250
            60
                      17640
            70
                      24010
            80
                      31360
            90
                      39690
```

Problem 2.15

In direct current applications, electrical power is calculated using Joule's law as

P = VI

where P is power in watts

V is the potential difference, measured in volts

I is the electrical current, measured in amperes

Joule's law can be combined with Ohm's law

V = IR

to give

$$P = I^2 R$$

where *R* is resistance measured in ohms.

The resistance of a conductor of uniform cross section (a wire or rod for example) is

$$R = \rho \frac{l}{A}$$

where

ho is the electrical resistivity measured in ohm-meters

I is the length of the wire

A is the cross-sectional area of the wire

This results in the equation for power

$$P = I^2 \rho \frac{l}{A}$$

Electrical resistivity is a material property that has been tabulated for many materials. For example

Material	Resistivity, ohm-meters (measured at 20°C)
Silver	1.59 × 10 ⁻⁸
Copper	1.68×10^{-8}
Gold	2.44×10^{-8}
Aluminum	2.82×10^{-8}
Iron	1.0×10^{-7}

rho =
$$[1.59e-8; 1.68e-8; 2.44e-8; 2.82e-8; 1.0e-7];$$
 % ohm-meters

Calculate the power that is dissipated through a wire with the following dimensions for each of the materials listed.

diameter 0.001 m

length 2.00 m

Assume the wire carries a current of 120 amps.

Problem 2.16

Repeat the previous problem for 10 wire lengths, from 1 m to 1 km. Use logarithmic spacing.

Realizing that 1km is 1000 meters, define a length array

```
length=logspace(1,3,10) % meters
length = 1 \times 10
           10
                    16.681
                                    27.826
                                                   46.416
                                                                 77.426
129.15 ...
P_silver=I.^2.*rho(1).*length./area
P_silver = 1 \times 10
        2915.2 4862.9
                                    8111.7
                                                    13531
                                                                  22571
37651 ...
P_copper=I.^2.*rho(2).*length./area
P copper = 1 \times 10
        3080.2 5138.1
                                    8570.9
                                                    14297
                                                                  23849
39783 ...
P_gold = I.^2.*rho(3).*length./area
P_gold = 1 \times 10
        4473.7 7462.5
                                     12448
                                                    20765
                                                                  34638
57779 ...
P_al = I.^2.*rho(4).*length./area
P al = 1 \times 10
```

```
5170.4 8624.7 14387 23999 40032

P_Fe = I.^2.*rho(5).*length./area

P_Fe = 1×10

18335 30584 51017 85102 1.4196e+05

2.368e+05 ...
```

In chapter 4 we'll learn how to do these calculations in a single step using the meshgrid function.

Problem 2.17

Newton's law of universal gravitation tells us that the force exerted by one particle on another is

$$F = G \frac{m_1 m_2}{r^2}$$

where the universal gravitational constant G is found experimentally to be

$$G = 6.673 \times 10^{-11} \text{ N m2/kg2}$$

The mass of each particle is m1 and m2, respectively, and r is the distance between the two particles. Use Newton's law of universal gravitation to find the force exerted by the earth on the moon, assuming that

the mass of the earth is approximately 6×10^{24} kg,

the mass of the moon is approximately 7.4 \times 10^{22} kg, and

the earth and the moon are an average of 3.9×10^8 m apart.

```
G=6.673e-11;  % N m^2/kg^2
m_earth=6e24;  % kg
m_moon=7.4e22;  % kg
r=3.9e8;  % meters
F=G*m_earth*m_moon/r^2
```

```
F = 1.9479e + 20
```

Note that the operators *, / and ^ were used because all of the values are scalars. Dot operators could also have been used and would give the same result.

Problem 2.18

We know that the earth and the moon are not always the same distance apart. Based on the equation in the previous problem, find the force the moon exerts on the earth for 10 distances between $3.8 \times 10^8\,$ m and $4.0 \times 10^8\,$ m. Be careful when you do the division to use the correct operator.

```
r=linspace(3.8e8,4.0e8,10); % meters
F=G*m_earth*m_moon./r.^2; % Notice the .^ and ./ operators
[r',F']
```

Recall from Problem 2.7 that the ideal gas law is:

PV = nRT

and that the Van der Waals modification of the ideal gas law is

$$\left(\left(P + \frac{n^2 a}{v^2}\right)(V - nb)\right) = nRT$$

Using the data from Problem 2.7, find the value of temperature (*T*), for

(a) 10 values of pressure from 0 bar to 400 bar for volume of 1 L

a)

```
P=linspace(0,400,10); % pressure
n=2; % moles
V=1; % volume
a=5.536; % constant
b=0.03049; % constant
R=0.08314472; % ideal gas constant
```

Find the temperature using the ideal gas law

```
T_ideal=P*V/(n*R)

T_ideal = 1×10

0 267.27 534.54 801.81 1069.1

1336.4 ...
```

Find the temperature using Van der Waal's equation

```
T_VW = (P+n^2*a/V^2)*(V-n*b)/(n*R)
T_VW = 1\times10
125.04 	 376.02 	 626.99 	 877.97 	 1128.9
1379.9 	 \cdots
```

It would be easier to compare the calculated values if they are listed in a table

```
[T_ideal',T_VW']
ans = 10 \times 2
                    125.04
            0
                    376.02
        267.27
       534.54
801.81
1069.1
1336.4
                    626.99
                    877.97
                    1128.9
       1336.4
                    1379.9
       1603.6 1630.9
1870.9 1881.9
                    2132.8
        2138.2
        2405.4 2383.8
```

(b) 10 values of volume from 0.1 L to 10 L for a pressure of 220 bar

```
V=linspace(0.1,10,10);
% Find the temperature using the ideal gas law
T_ideal=P*V/(n*R)
T_ideal = 1 \times 10
        132.3
                   1587.6
                                3042.9
                                               4498.2
                                                            5953.5
7408.8 ...
% Find the temperature using Van der Waal's equation
T_VW=(P+n^2*a./V.^2).*(V-n*b)/(n*R)
T VW = 1 \times 10
       571.23 1612.2 3018.6
                                                 4456
                                                              5902
7351.6 ...
% It would be easier to compare the calculated values if they are listed in
[T_ideal',T_VW']
ans = 10 \times 2
        132.3
                   571.23
       1587.6
                   1612.2
                   3018.6
       3042.9
                    4456
5902
       4498.2
       5953.5
       7408.8 7351.6
```

8864.1	8803.1
10319	10256
11775	11709
13230	13163

C

You might interpret this probem to mean that the 10 values of pressure

and ten values of volume should be used in the same calculation

```
V=linspace(0.1,10,10);
P=linspace(0,400,10);
% Find the temperature using the ideal gas law
T ideal=P.*V/(n*R)
T ideal = 1 \times 10
                 320.73 1229.4
           0
                                               2726.2
                                                             4810.9
7483.6 ...
% Find the temperature using Van der Waal's equation
T_VW=(P+n^2*a./V.^2).*(V-n*b)/(n*R)
T VW = 1 \times 10
       519.61 409.76 1253.2
                                               2715.7
                                                             4774.9
7425.6 ...
% It would be easier to compare the calculated values if they are listed in
% a table
[T_ideal',T_VW']
ans = 10 \times 2
                  519.61
            0
       320.73
                   409.76
       1229.4 1253.2
2726.2 2715.7
4810.9 4774.9
                  7425.6
       7483.6
        10744
                   10666
                     14496
        14593
        19030
                     18914
         24054
                     23921
```

Number Display

Problem 2.20

Create a array **a** equal to [-1/3, 0, 1/3, 2/3], and use each of the built-in format options to display the results:

```
format short (which is the default)
format long
```

```
format bank
   format short e
   format long e
   format short eng
   format long eng
   format short g (shortg)
   format long g (longg)
   format +
   format rat
 a=[-1/3,0,1/3,2/3] % displays as format shortg automatically since shortg was
specified at the beginning of this file
 a = 1 \times 4
                                     0.33333 0.66667
       -0.33333
                             0
format long
а
 a = 1 \times 4
   -0.333333333333333
                                                 0.33333333333333 ...
format bank
а
 a = 1 \times 4
           -0.33
                                0
                                            0.33
                                                            0.67
format short e
 a = 1 \times 4
   -3.3333e-01
                                  3.3333e-01
                                                 6.6667e-01
format long e
 a = 1 \times 4
     -3.3333333333333e-01
                                                           0
 3.3333333333333e-01 ···
format short eng
 а
```

```
a = 1 \times 4
   -333.333e-003 0.0000e+000 333.333e-003 666.6667e-003
format long eng
а
a = 1 \times 4
    -333.33333333333e-003 0.000000000000e+000
                                                            333.3333333333e-003
format short
а
a = 1 \times 4
    -0.3333
                     0 0.3333 0.6667
format long g
а
a = 1 \times 4
         -0.333333333333333
                                                         0
0.33333333333333 ...
format +
a
a = 1 \times 4
- ++
format rat
а
                        0
a =
         -1/3
                                      1/3
                                                   2/3
format shortg % This is my favorite
а
a = 1 \times 4
      -0.33333
                                    0.33333 0.66667
```

Saving Your Work in Files

Problem 2.21

Create an array called D_to_R composed of two columns, one representing degrees, and the other representing the corresponding value in radians. Any value set will do for this exercise.

- Save the array to a file called degrees.dat.
- Once the file is saved, clear your workspace then load the data from the file back into MATLAB.

```
D=0:10:180;
```

```
R=D*pi/180;
D_to_R=[D',R']
D_{to}R = 19 \times 2
                             0
             10
                      0.17453
             20
                      0.34907
                       0.5236
             30
                      0.69813
             40
                      0.87266
             50
             60
                       1.0472
             70
                       1.2217
                       1.3963
             80
             90
                       1.5708
save degrees.dat -ascii D_to_R
```

Check your current directory to confirm that the file was saved

```
clear
load degrees.dat
degrees
degrees = 19 \times 2
              0
                             0
             10
                      0.17453
                      0.34907
             20
             30
                       0.5236
             40
                      0.69813
             50
                      0.87266
                       1.0472
             60
                       1.2217
             70
                       1.3963
             80
             90
                       1.5708
```